Drexel University University
CHEM 425: Inorganic Chemistry Laboratory
Spring 2014 Syllabus

INSTRUCTORS

• Dr. Molly A. O’Connor (course coordinator)
  o Office: Stratton 410
  o Email: mao23@drexel.edu (also maoconnor83@gmail.com)
  o Phone: (215) 895-2666
  o Office Hours: Tues and Fri 10 am – 12 pm in Stratton 410; also available by appointment

• Mr. Mohammad Nozari
  o Office: Disqué 402
  o Email: mn468@drexel.edu
  o Phone: (215) 895-1697
  o Office Hours: Wed 2 – 3 pm in Disqué 402; also available by appointment

• Mr. Kyle Hess
  o Office: Disqué 511
  o Email: krh52@drexel.edu
  o Phone: (215) 895-1806
  o Office Hours: available by appointment

• Dr. Curtis Wahlgren
  o Email: cgw24@drexel.edu
  o Office Hours: On Campus M & F 12 – 3 pm and W 2 – 5 PM; also available by appointment

CLASS SCHEDULE

• Lab Lecture (All sections): Monday 10:00 – 10:50 am Stratton 219
• Lab (Section 061): Wednesday 11:00 am – 1:50 pm Disqué 409
  Friday 1:00 – 3:50 pm Disqué 409
• Lab (Section 062): Tuesday 2:00 – 4:50 pm Disqué 409
  Friday 5:00 – 7:50 pm Disqué 409

COURSE OBJECTIVES

• To improve your skills in the handling of chemical reagents.
• To learn about some types of reactions that are new to you.
• To apply and experience principles learned in the Inorganic lectures.
• To increase your skills in the interpretation of experimental results.

Note – CHEM 422 is a pre or co-requisite for CHEM 425 (the inverse does not apply).

COURSE MATERIALS

• Lab Experiments will be provided for students during the course.
• Eye protection and lab coat are required during the lab period.
• Additional laboratory texts can be found in Hagerty Reserves.
COURSE EXPECTATIONS

Lab: The lab sequence involves (a) regularly scheduled experiments performed by everybody in the class, plus (b) an "individual project", chosen and performed by one pair of students. Your grade for the course will be determined according to your laboratory reports. Keep in mind that a lab report should communicate not only the quantitative results of the experiment but also the quality of your data and your understanding of the experiment. Reports should be done using the appropriate software and are typically due 2 weeks after the completion of the experiment (see tentative course schedule). If a report is submitted late, then for each Drexel class day by which a report is overdue, **5% of the marks are deducted**. In any case, any and all lab reports for a grade **must be submitted no later than 5:00 PM, Mon. June 9th**. Your Chem 425 grade is also dependent on timely attendance at the lab-lecture, and participation in the attendant discussion. In general, by far the major part of your grade is derived from doing the experiments and writing good lab reports that are submitted on time. You may choose to perform experiments in "teams" of two people. However, the lab reports for the experiment must be written independently of the other team member, even though the data would be utilized in common. The lowest lab grade will be dropped for each student, excluding the individual project grade.

Lab reports: The format of the reports should be that of a research paper in ACS Inorganic Chemistry. Be sure to include the following elements: (1) Title Page; (2) Introduction (be concise); (3) Experimental Procedure; (4) Experimental Data (tabular format, with proper units); (5) Results & Conclusions; (6) References. In (7) an Appendix: show the calculations used in obtaining intermediate and final results, including any equations used in proceeding from data to results. In case of duplicate, triplicate, or other multiple sets of data, only one set of calculations need be shown, but all results should be listed in tabular form. When averaging or data-fitting is called for, a standard deviation should be listed as well as a mean. Also include any graphs, charts, and spectra from the experiment. Answers to any specific questions asked within the experiment are typically incorporated into the applicable sections of the lab report, though it is very helpful to the instructors to include a section in the Appendix stating where the answers to the questions can be found.

Regularly Scheduled Experiments: The following experiments will be performed during the term (also see the tentative course schedule)

- Crystal Field Properties of Ni$^{2+}$/Cu$^{2+}$ Compounds.
- Synthesis and Excited-State Chemistry of Ru(Bipy)$_3^{2+}$
- Transport across a Model Liquid Membrane
- Assembly and Properties of a Macrocyclic Complex.
- Synthesis of a Silicone Polymer
- An Inorganic Dye-Based Thin-film Photovoltaic Cell.
- (Arene)Mo(CO)$_3\pi$-Complexes
- A Ruthenium Nitrosyl Complex from Diazald®
- Ligation Equilibrium of a Metalloprotein

Individual Projects:
You and your lab partner will choose one of the provided projects from the instructor's sign-up sheet, and initial that experiment to claim it. The list of experiments will be available in week 2. You may also wish to design your own experiment, pending instructor approval. The tentative schedule with regards to individual projects is as follows:

- By April 11th: sign up for your choice of project.
- By April 18th: submit your written proposal for performing the project.
  - You will need to outline: (1) Apparatus and (amounts of) reagents required; (2) The experimental procedure; (3) A description of the methods to be used for appropriately characterizing the product(s), indicating what measurements will be made; (4) your planned timetable for
performing the work; (5) References. The Instructors will review and critique your proposal.

It’s obviously necessary to allow some lead time for purchasing any required reagents.

• You will have your second lab period of week 7 plus both your lab periods of week 8 (May 19-23) to perform the experimental work. There is also free time (e.g., while refluxing a reaction) during earlier experimental periods, when you can get started on your project.

• June 2 - June 8: write up your experiment. The written report should again be in the format of a research paper in ACS Inorganic Chemistry. **Note – only 1 lab report need be submitted per group.**

**Web:** The “BbLearn” course website will be used throughout the course. Lab lecture notes, lab materials, and course announcements will be regularly posted. Emails will also be sent to your Drexel email account. Students are responsible for checking the course website and email on a regular basis.

**Other University Policies:**

• Graduating Seniors might wish to review the university policies on Final Examinations and Senior Privilege: [http://www.drexel.edu/provost/policies/examinations.html](http://www.drexel.edu/provost/policies/examinations.html)

• Disability accommodations: [http://www.drexel.edu/oed/disabilityResources/](http://www.drexel.edu/oed/disabilityResources/)

• Dropping courses: [http://www.drexel.edu/provost/policies/course_drop.asp](http://www.drexel.edu/provost/policies/course_drop.asp)

**Broad statement of the goals of 425 in the B.S. program context; development of abilities to:**

• write scientific reports, with graphical presentation of data (technical writing skills)

• document scientific information and experimental data (keeping a lab notebook or other written record)

• use theory to understand/predict experimental observations

• make quantitative/structural measurements and interpret results (data collection and analysis)

• identify and handle hazardous materials (includes ability to access this information and to apply reasonable cautionary measures)

• plan and execute chemical transformations (experimental skills)

• select and use chemical instrumentation and spectroscopy

• work in teams and develop collaborations

• use the physical principles upon which chemical instrumentation is based

• understand and use reaction chemistry

• find and evaluate the validity and usefulness of information (e.g., library or reputable online resources)

• discern and practice good scientific ethics (e.g., avoiding plagiarism)

• discern sources of experimental error

• use statistics to judge limitations of data

• use math and logic to understand data and results

**GRADING**

Tentative grade cutoffs are shown below. The instructors reserve the right to lower some or all of the grade cutoffs.

<table>
<thead>
<tr>
<th>Final Grade</th>
<th>A</th>
<th>A-</th>
<th>B+</th>
<th>B</th>
<th>B-</th>
<th>C+</th>
<th>C</th>
<th>C-</th>
<th>D+</th>
<th>D</th>
<th>F</th>
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<tbody>
<tr>
<td>%</td>
<td>100-93</td>
<td>92-90</td>
<td>89-87</td>
<td>86-83</td>
<td>82-80</td>
<td>79-77</td>
<td>76-73</td>
<td>72-70</td>
<td>69-67</td>
<td>66-65</td>
<td>Below 65</td>
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**Academic Integrity:** Discussion of lecture material and homework assignments among class members is encouraged, but all final lab reports turned in for grading must be your own work, unless otherwise explicitly stated. Academic dishonesty in any form (plagiarism, cheating, copying lab reports, etc.) will not be tolerated and will result in an **AUTOMATIC ZERO** on the given assignment. Subsequent incidents of academic dishonesty will result in automatic failure of the course. For more information, see material in “academic dishonesty” under the “academic policies” tab at the following link: [http://drexel.edu/studentaffairs/community_standards/studentHandbook/](http://drexel.edu/studentaffairs/community_standards/studentHandbook/)
TENTATIVE COURSE SCHEDULE

The instructors reserve the right to change the schedule of lab experiments and lab report due dates, if necessary. Appropriate advance notice will be given by in-class announcement and through the course website.

<table>
<thead>
<tr>
<th>Week</th>
<th>Laboratory Experiment</th>
<th>Notes and Due Dates</th>
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| 1 (3/31 – 4/4) | **Lab Day 1**: Crystal Field Properties of Ni\(^{2+}\)/Cu\(^{2+}\) Compounds  
**Lab Day 2**: Crystal Field Properties of Ni\(^{2+}\)/Cu\(^{2+}\) Compounds | Sign-up for your choice of Individual Project on Lab Day 2 |
| 2 (4/7 – 4/11) | **Lab Day 1**: Synthesis and Excited-State Chemistry of Ru(Bipy)\(_3\)^{2+}  
**Lab Day 2**: Synthesis and Excited-State Chemistry of Ru(Bipy)\(_3\)^{2+} | Ni\(^{2+}\)/Cu\(^{2+}\) Compounds lab report due on Lab Day 2 |
| 3 (4/14 – 4/18) | **Lab Day 1**: Transport across a Model Liquid Membrane and ligand synthesis from “Assembly and Properties of a Macrocyclic Complex”  
**Lab Day 2**: Assembly and Properties of a Macrocyclic Complex | Submit your written proposal for Individual Project on Lab Day 2 |
| 4 (4/21 – 4/25) | **Lab Day 1**: Assembly and Properties of a Macrocyclic Complex  
**Lab Day 2**: Synthesis of a Silicone Polymer | Ru(Bipy)\(_3\)^{2+} lab report due on Lab Day 2 |
| 5 (4/28 – 5/2) | **Lab Day 1**: An Inorganic Dye-Based Thin-film Photovoltaic Cell  
**Lab Day 2**: An Inorganic Dye-Based Thin-film Photovoltaic Cell | Membrane lab report due on Lab Day 1 |
| 6 (5/5 – 5/9) | **Lab Day 1**: (Arene)Mo(CO)\(_3\) \(\pi\)-Complexes and A Ruthenium Nitrosyl Complex from Diazald\(^\text{®}\)  
**Lab Day 2**: (Arene)Mo(CO)\(_3\) \(\pi\)-Complexes and A Ruthenium Nitrosyl Complex from Diazald\(^\text{®}\) | Ni Macro lab report due on Lab Day 1  
Silly Putty lab report due on Lab Day 2 |
| 7 (5/12 – 5/16) | **Lab Day 1**: (Arene)Mo(CO)\(_3\) \(\pi\)-Complexes and A Ruthenium Nitrosyl Complex from Diazald\(^\text{®}\)  
**Lab Day 2**: Individual Projects | Photovoltaic Cell lab report due on Lab Day 2 |
| 8 (5/19 – 5/23) | **Lab Day 1**: Individual Projects  
**Lab Day 2**: Individual Projects |                                |
| 9 (5/26 – 5/30) | **Lab Day 1**: Ligation Equilibrium of a Metalloprotein  
**Lab Day 2**: Additional Individual Project time if needed | No Classes Monday May 26\(^\text{th}\)  
(Arene)Mo(CO)\(_3\) lab report due Lab Day 1 |
| 10 (6/2 – 6/6) | **Lab Day 1**: Additional Individual Project time if needed  
**Lab Day 2**: Additional Individual Project time if needed | Ruthenium Nitrosyl lab report due Lab Day 1 |
| 11 (6/9 – 6/13) | **Laboratory Drawer Check out During Lab Lecture Monday 6/9**  
**Final Exams 6/10 – 6/13** | Metalloprotein lab report due Monday June 9\(^\text{th}\)  
Individual Project lab report due Monday June 9\(^\text{th}\)  
ALL LAB REPORTS must be submitted by 5 pm on Monday June 9\(^\text{th}\)! |